

3-[3-(4-Ethoxyphenyl)-1-phenyl-1*H*-pyrazol-4-yl]-
4-nitro-2,5-diphenylisoxazolidineD. Gayathri,^a D. Velmurugan,^{a*}
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Key indicators

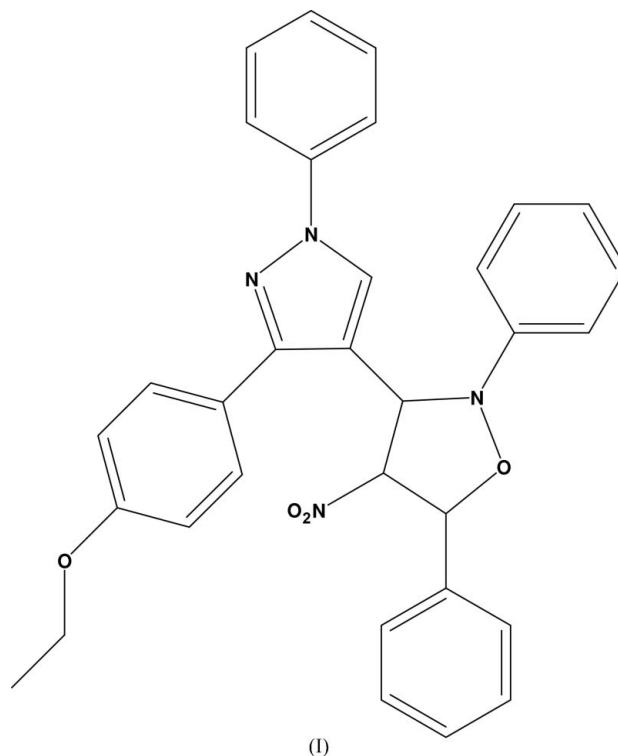
Single-crystal X-ray study
T = 273 K
Mean $\sigma(\text{C}-\text{C})$ = 0.003 Å
R factor = 0.051
wR factor = 0.144
Data-to-parameter ratio = 17.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.The isoxazolidine ring in the title compound, $\text{C}_{32}\text{H}_{28}\text{N}_4\text{O}_4$,
adopts an envelope conformation.

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Comment

Isoxazolidines are interesting intermediates for the synthesis of several classes of natural products and biologically active compounds such as β -amino acids and alkaloids (Gothelf & Jorgensen, 1998). Many pyrazole derivatives are known to exhibit a wide range of biological properties such as anti-hyperglycemic, analgesic, anti-inflammatory, antipyretic, anti-bacterial, hypoglycemic, sedative-hypnotic and anticoagulant activities. In particular, arylpyrazoles are widely used in medicinal and pesticidal chemistry. Recently, some arylpyrazoles were reported to display non-nucleoside HIV-1 reverse transcriptase inhibitory activity (Lee *et al.*, 2003).



Bond lengths and angles in the title compound, (I), are comparable with literature values (Allen *et al.*, 1987). The sum of the bond angles around N3 [332.1°] indicates sp^3 hybridisation. The pyrazole ring forms dihedral angles of 20.7 (1)° with the phenyl ring attached at N1 and 51.4 (1)° with the 4-ethoxyphenyl ring at C13. The isoxazolidine ring adopts a twisted conformation with a pseudo-twofold axis passing through atom C19 and bond O4—N3.

Experimental

A solution of pyrazolyl nitron (0.5 mmol) and β -nitrostyrene (0.5 mmol) was refluxed in dry toluene (10 ml). The completion of the reaction was monitored by thin-layer chromatography. The solvent was evaporated under reduced pressure to give the crude product, which was further purified by column chromatography using petroleum ether–ethyl acetate (96:4) as eluent, to afford the pure product (68%) as a pale-yellow solid. Single crystals were obtained by crystallization from chloroform.

Crystal data

$C_{32}H_{28}N_4O_4$	$V = 2720.8 (3) \text{ \AA}^3$
$M_r = 532.58$	$Z = 4$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 13.2463 (7) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$b = 10.0112 (6) \text{ \AA}$	$T = 273 (2) \text{ K}$
$c = 20.9518 (12) \text{ \AA}$	$0.27 \times 0.24 \times 0.23 \text{ mm}$
$\beta = 101.688 (1)^\circ$	

Data collection

Bruker SMART CCD area-detector diffractometer	6286 independent reflections
Absorption correction: none	4903 reflections with $I > 2\sigma(I)$
30188 measured reflections	$R_{int} = 0.024$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.051$	362 parameters
$wR(F^2) = 0.144$	H-atom parameters constrained
$S = 1.03$	$\Delta\rho_{max} = 0.34 \text{ e \AA}^{-3}$
6286 reflections	$\Delta\rho_{min} = -0.18 \text{ e \AA}^{-3}$

All H atoms were refined using a riding model, with C–H = 0.93 \AA , $U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic, C–H = 0.98 \AA , $U_{iso}(H) = 1.2U_{eq}(C)$ for CH, C–H = 0.97 \AA , $U_{iso}(H) = 1.2U_{eq}(C)$ for CH_2 , C–H = 0.96 \AA , $U_{iso}(H) = 1.5U_{eq}(C)$ for CH_3 atoms and N–H = 0.86 \AA , $U_{iso}(H) = 1.2U_{eq}(N)$ for the NH group.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics:

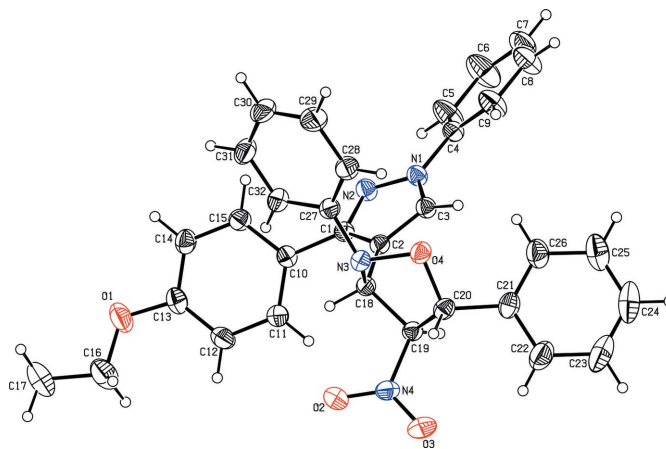


Figure 1
The molecular structure of the title compound, showing 30% probability displacement ellipsoids.

PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97 and PARST (Nardelli, 1995).

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